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#### **Key indicators**

Single-crystal X-ray study T = 299 K Mean  $\sigma$ (C–C) = 0.006 Å Disorder in main residue R factor = 0.057 wR factor = 0.152 Data-to-parameter ratio = 13.9

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# 2-Aminoquinolin-4-yl 2,4,6-triisopropylbenzenesulfonate

The molecular packing of the title compound,  $C_{24}H_{30}N_2O_3S$ , is stabilized by a hydrogen-bonded network. Both sulfonyl O atoms are involved in intermolecular hydrogen bonds of types  $N-H\cdots O$  and  $C-H\cdots O$ . An intermolecular  $N-H\cdots N$  hydrogen bond is also observed.

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### Comment

Quinolines comprise an important class of heterocyclic compounds present in many potent biologically active molecules (Frank et al., 2004). In addition, aminoquinoline derivatives have shown cytotoxic activity (Kim et al., 2005), as well as antibacterial, antifungal and antiparasitic activities (Jain et al., 2005). A large number of quinoline derivatives have been synthesized because the quinoline unit has well defined and attractive ionophoric properties toward a variety of important metal ions (Yoshida et al., 2002). The specific properties of the quinoline fluorophores that exhibit dramatic fluorescence enhancement upon complexation with guest molecules will be useful for fundamental research into solid-state fluorescence and for the development of new intense solid-emissive materials (Ooyama et al., 2005). Our interest in such metal chelators is as potential agents for neuroprotection in neurodegenerative diseases (Zheng et al., 2005); in connection with this research, we describe in this paper the crystallographic study of the title compound, (I).



The quinoline ring system, with the amino group, is nearly planar, with maximum deviations from the mean plane of -0.032 (3) Å for atom N1 and 0.042 (3) Å for atom N2. The quinoline unit forms a C1-O1-S1-C10 torsion angle with the benzene ring of 87.7 (2)°. Three intermolecular hydrogen bonds of types N $-H\cdots$ O, N $-H\cdots$ N and C $-H\cdots$ O are observed; these connect molecules to form a three-dimensional network (Fig. 2). Details of the hydrogen-bonding parameters are given in Table 1.

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## Experimental

Compound (I) was prepared according to a literature procedure (Xue *et al.*, 2000). Crystals suitable for X-ray diffraction analysis were obtained by recrystallization from a methanol–dichloromethane solution (1:1).

V = 1162.2 (3) Å<sup>3</sup>

 $D_x = 1.219 \text{ Mg m}^{-3}$ 

 $0.28 \times 0.13 \times 0.08 \text{ mm}$ 

frequency: 120 min

intensity decay: 2.5%

H atoms treated by a mixture of

 $w = 1/[\sigma^2(F_o^2) + (0.0729P)^2]$ where  $P = (F_o^2 + 2F_c^2)/3$  $(\Delta/\sigma)_{\text{max}} = 0.012$ 

independent and constrained

Cu  $K\alpha$  radiation

 $\mu = 1.45 \text{ mm}^{-1}$ 

T = 299 (2) K Prism, yellow

 $R_{\rm int} = 0.027$ 

 $\theta_{\rm max} = 66.9^{\circ}$ 3 standard reflections

refinement

 $\Delta \rho_{\text{max}} = 0.23 \text{ e } \text{\AA}^{-3}$  $\Delta \rho_{\text{min}} = -0.33 \text{ e } \text{\AA}^{-3}$ 

Z = 2

#### Crystal data

 $\begin{array}{l} C_{24}H_{30}N_2O_3S\\ M_r = 426.56\\ \text{Triclinic, } P\overline{1}\\ a = 9.431 \ (1) \ \mathring{A}\\ b = 11.015 \ (2) \ \mathring{A}\\ c = 12.312 \ (2) \ \mathring{A}\\ \alpha = 77.61 \ (1)^\circ\\ \beta = 80.79 \ (1)^\circ\\ \gamma = 69.11 \ (1)^\circ \end{array}$ 

#### Data collection

Nonius CAD-4 diffractometer  $\omega/2\theta$  scans Absorption correction: none 4540 measured reflections 3997 independent reflections 2461 reflections with  $I > 2\sigma(I)$ 

#### Refinement

Refinement on  $F^2$   $R[F^2 > 2\sigma(F^2)] = 0.057$   $wR(F^2) = 0.152$  S = 1.023997 reflections 287 parameters

## Table 1

Hydrogen-bond geometry (Å, °).

$D - \mathbf{H} \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
N2-H21 $N$ ···O3 <sup>i</sup> N2 H22 $N$ ···N1 <sup>ii</sup>	0.84(4) 0.85(4)	2.37(4)	3.197 (4) 3.073 (4)	167 (4) 175 (4)
$C9-H9\cdots O2^{i}$	0.93	2.60	3.522 (4)	173 (4)

Symmetry codes: (i) -x, -y + 1, -z + 1; (ii) -x - 1, -y + 2, -z + 1.

The amino H atoms were located in a difference map and refined freely. Carbon-bound H atoms were positioned with idealized geometry using a riding model (C-H = 0.93-0.98 Å). All H atoms were refined with isotropic displacement parameters (set at 1.2 times of the  $U_{eq}$  of the parent atom). Atom C20 of the methyl group is disordered and was refined with a split model. The corresponding site-occupation factors were refined, but were later fixed at 0.65:0.35.

Data collection: *CAD-4-PC* (Nonius, 1996); cell refinement: *CAD-4-PC*; data reduction: *REDU4* (Stoe & Cie, 1987); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97*.

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#### Figure 1

The molecular structure of (I), showing the atom labeling and displacement ellipsoids drawn at the 50% probability level. The minor disorder components are shown with dashed bonds.



#### Figure 2

The molecular packing of (I), with hydrogen bonds shown as dashed lines.

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